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6 Proton Magnetic Resonance Spectroscopy Studies of
Aluminum (III), Gallium (III), and Indium(III) in
Methanol and Ethanol: Determination of Solvation
Numbers and Exchange Rates.

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Dr. David Richardson

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The objectives of this research endeavor are to determine the solvation numbers and rates of ligand exchange for selected aluminum, gallium, and indium salts in methanol and ethanol utilizing a Varian EM-390 Nuclear Magnetic Resonance Spectrometer. ↗

Special sample preparation techniques were employed to make the anhydrous inorganic salts. These methods were similar to those used by Donoghue and Drago¹ to prepare anhydrous cobalt perchlorate. Anhydrous aluminum nitrate in ethanol, aluminum perchlorate in ethanol, gallium perchlorate in ethanol, indium chloride in methanol and ethanol, indium nitrate in methanol and ethanol and indium perchlorate in methanol and ethanol have prepared via the chemical dehydration of their respective hydrated salts with 2,2-dimethoxypropane. The order of addition of reactants is important. If the hydrated salt is first added to 2,2-dimethoxypropane or 2,2-diethoxypropane, a strong brown discoloration is produced. However, if the hydrated salt is added to "super-dry" methanol or ethanol, and then the 2,2-dialkoxypropane the discoloration is reduced. Nevertheless, in the case of indium salts some discoloration results regardless of the order of addition.



The "super dry" alcohols were prepared by treatment with magnesium turnings using iodine to initiate the reaction as described by Vogel². The alcohols were stored over Linde 4A molecular sieve in a nitrogen environment.

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All starting materials and other chemicals were reagent grade quality. Solutions for nmr measurements were prepared by adding a specific quantity of anhydrous alcohol to the appropriate anhydrous salt in a dry environment (glove bag under a nitrogen atmosphere). The measured anhydrous mixtures were accurately weighed. The samples were sealed in standard nmr tubes following freeze-thaw degassing, and stored in a dry ice-2-propanol mixture when not in use. Samples used for solvation number determinations were approximately 0.5 M, while those used for rate determinations were approximately 0.25 M.

Measurements for solvation numbers and exchange rates were recorded on a Varian EM-390 spectrometer equipped with a special temperature controller. Signal areas were integrated with an Alvin PL-655M planimeter.

Ligand-exchange rates were obtained by curve-fitting experimental OH slow-passage spectra with computer calculated spectra for several trial exchange rates, as well as by the Bloch equation for the region for which the slow exchange approximation is valid (Appendix A). Measurements of parameters required for rate calculations were made over a wide range of temperatures.

Consultations of methodologies described and interpretation of data obtained up-to-date were made with Dr. Terry D. Alger and Dr. Joseph Morse (Higher Board of Education of Utah and Utah State University respectively).

Spectra of $\text{In}(\text{ClO}_4)_3$ in ethanol were taken from -30°C to -70°C for solvation studies and spectra for rate studies were taken from -5°C to -95°C .

Spectra for InCl_3 in ethanol were taken from -65°C to -90°C for solvation studies and -60°C to -95°C for rate studies.

The rate studies and the solvation studies for $\text{In}(\text{ClO}_4)_3$ in ethanol proved to be of particular significance. The result of the rate studies is indicated in figure 1. The computer program has recently been debugged, therefore, the thermodynamic parameters sought have not as of yet been ascertained. An approximation of what is anticipated and what will be used to calculate the thermodynamic parameters (which will be used to predict the mechanism of ligand exchange) is indicated in figure 2.

The solvation number determination proved to be fantastically interesting (Table I). There is a definite change in the solvation number (never before observed in such a dramatic manner) as the temperature is lowered. Evidence strongly suggest that the solvation effects undergo a rather remarkable modification as the temperature is lowered for the indium system in ethanol. The spectra for this system will have to be photographed, therefore, they will be included in the final report. Nevertheless, such a system apparently exhibits a coordination complex of seven at lower temperatures and a coordination of six at higher temperatures. Further discussion of this will appear in the final report.

Additional spectra and other systems have not been interpreted or obtained since the EM-390 has been made fully operational only very recently.

Serious difficulties have been encountered with this research activity. The magnet to the EM-390 was dropped by individuals employed to move it to a specified location. The magnet had to undergo extensive repairs. This unfortunate accident set the project back several months. The instrument has recently been properly installed and is functioning at peak efficiency.

Considerable problems were encountered in preparing $\text{In}(\text{ClO}_4)_3$ in methanol. The samples gave rather broad nmr peaks which were not ideal for solvation number determinations.

Problems were also obtained in debugging the many site nmr Line Shape Program.

$\log W_{1/2}$

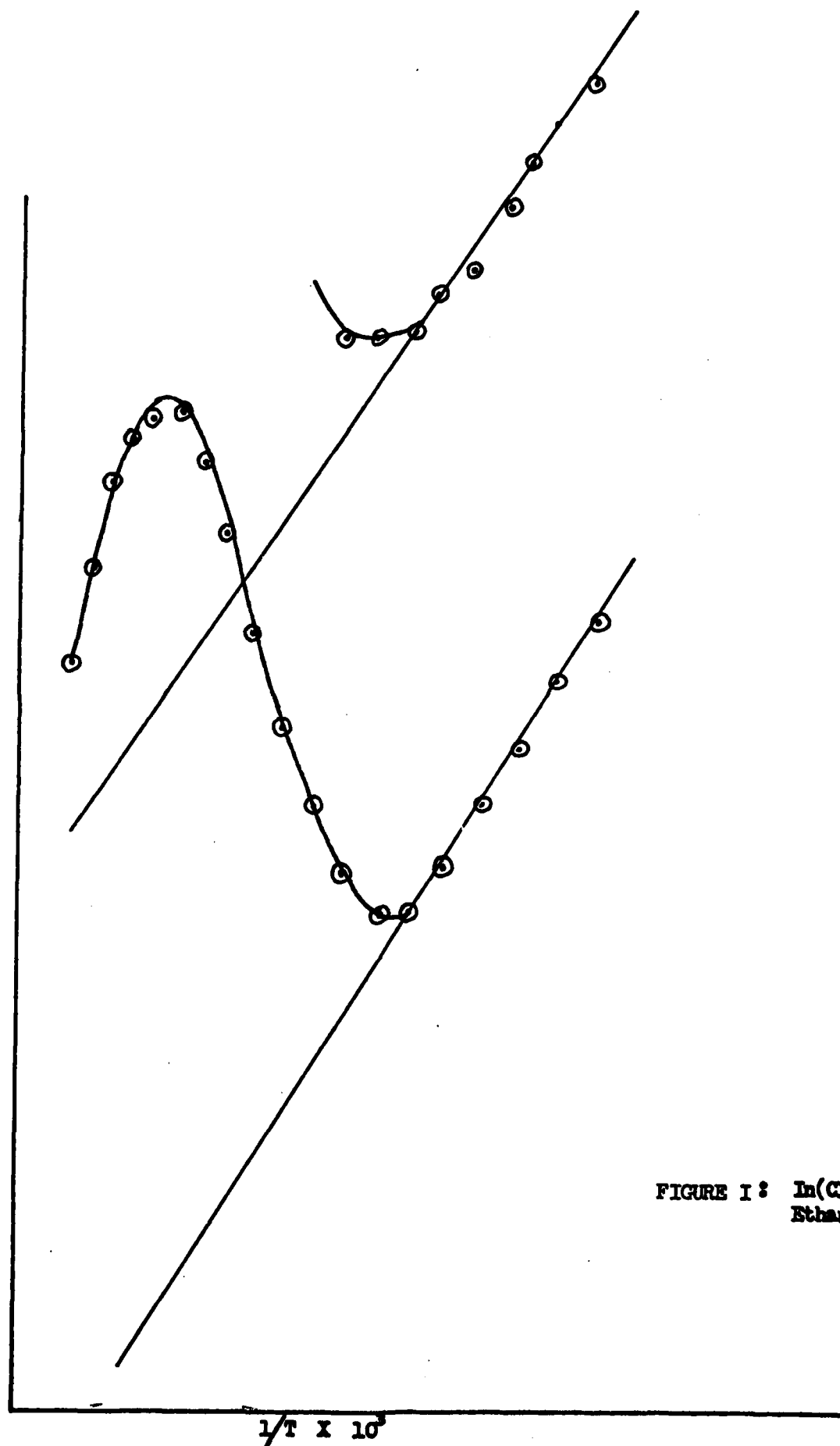


FIGURE I: $\text{In}(\text{ClO}_4)_3$ in Ethanol

log k

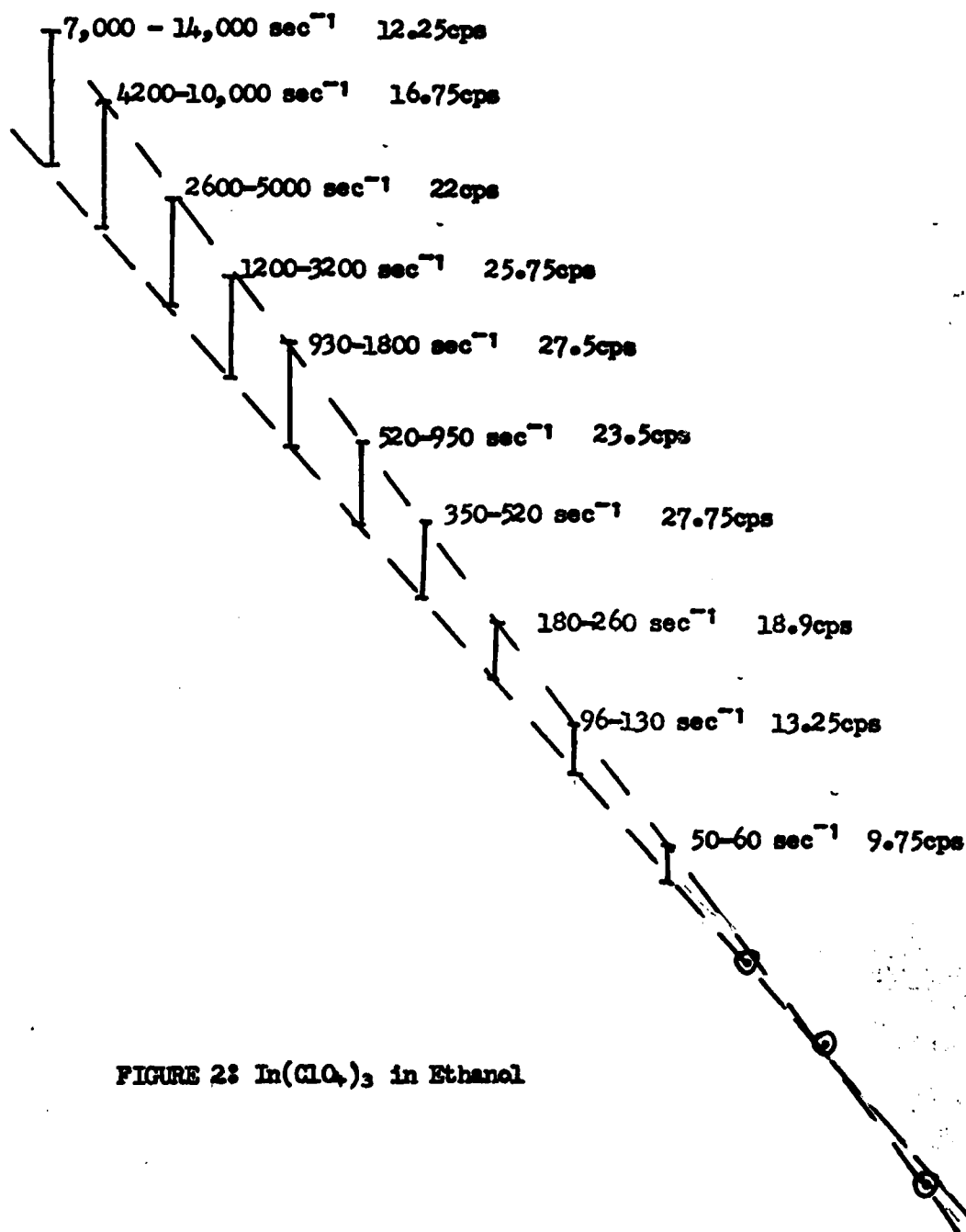


Table I: Determination of the Solvation Number — In(III) in Ethanol

Temp °C	moles of Indium Perchlorate $\times 10^3$	Molar Ratio M/M	Solvation OH A_s	Area under Bulk OH peak	Solvation Number
-40	5.1	33.33	0.057	0.131	5.9
-40	5.1	33.33	0.049	0.220	6.07
-45	5.1	33.33	0.040	0.196	5.64
-50	5.1	33.33	0.030	0.116	6.8
-50	5.1	33.33	0.034	0.131	6.9
-55	5.1	33.33	0.026	0.1053	6.6
-60	5.1	33.33	0.026	0.0985	6.96
-70	5.1	33.33	0.026	0.095	7.2

Reference

1. J. T. Donoghue and R. S. Drago, *Inorg. Chem.*, 1, 866(1962)
2. Arthur I. Vogel: *A Textbook of Practical Organic Chemistry*, third edition, page 166.
Longman

APPENDIX A

SAUNDERS MANY SITE NMR LINE-SHAPE PROGRAM
PROGRAM FOR CALCULATING NMR LINE SHAPES FOR CASES INVOLVING UP TO 25 LINES. ANY PROBABILITIES FOR TRANSITIONS BETWEEN THE VARIOUS SITES MAY BE INCLUDED, BUT SPIN-SPIN COUPLING SHOULD BE ABSENT UNLESS IT IS FIRST ORDER IN WHICH CASE IT MAY BE CONSIDERED TO PRODUCE ADDITIONAL SITES IF DESIRED, THE SYSTEM PRINTER MAY BE USED TO PLOT THE LINE-SHAPE.

THE PROGRAM FIRST READS A CONTROL CARD WITH THE FOLLOWING INFORMATION

N(1) # ORDER OF MATRIX (THE NUMBER OF SITES)

N(2) WHEN POSITIVE CALLS PLOT WHICH DRAWS A GRAPH ON THE SYSTEM PRINTER

N(3) POSITIVE SUPPRESSES PRINTING OF SPECTRUM

N(4) WHEN POSITIVE PROGRAM OMITS READING NEW PROBABILITY MATRIX AND GOES ON TO READ A NEW RATE CARD.

N(5) WHEN EQUAL TO 1, CHECKS TRANSITION PROBABILITY MATRIX AGAINST EQUILLIBRIUM PROBABILITIES AND EXITS IF AN ERROR IS FOUND.

N(7) WHEN POSITIVE ADDS SPECTRUM TO PREVIOUS ONE. NO RATE CARD IS NEEDED THEN A CARD WITH THE FREQUENCIES (IN CPS), THE RELATIVE EQUILLIBRIUM POPULATIONS, THE HALF WIDTHS OF THE LINES IN THE ABSENCE OF EXCHANGE, THE RELATIVE EQUILLIBRIUM POPULATIONS, AND THE NUMBER OF SITES TO WHICH THE PARTICULAR LINE IN QUESTION CAN GO IS READ.

THE PROGRAM THEN READS CARDS WITH THE RELATIVE PROBABILITIES FOR GOING FROM THE I'TH SITE TO THE M'TH SITE.

AFTER THE COMPLETE PROBABILITY MATRIX IS READ, THE PROGRAM READS A CARD WITH THE RATE CONSTANT(PSEUDO FIRST ORDER IN 1/SEC),THE FIRST AND LAST FREQUENCIES AT WHICH THE INTENSITY IS TO BE CALCULATED AND THE INTERVAL FOR THIS CALCULATION.

ONE CAN CONTINUE WITH A CONTROL CARD WITH A ONE IN COLUMN 20 FOLLOWED BY ANOTHER RATE CARD ETC. IN ORDER TO REPEAT THE CALCULATION AT DIFFERENT RATES.

DIMENSION N(7), W(25), P(25), TR(25), SR(2000), A(25,25), Q(25)

DIMENSION T2(25), B(25,25), FG(2000)

DIMENSION C(25,25), D(25,25)

COMMON N

READ CONTROL CARD

```
1 READ (5,5)N
5 FORMAT(14I5)
WRITE (6,8)N
FORMAT (1H1, 5X, 14I7)
IF(N(7).GE.1) GO TO 9
DO 11 NN = 1,2000
11 SR(NN) = 0.
NA = N(1)
IF(NA.GT.25.OR.NA.LE.0) GO TO 1
6 IF(N(4)-1)2,27,1
```

CLEAR PROBABILITY MATRIX

```
DO 30 I = 1, NA
DO 30 J = 1, NA
B(I, J) = 0.
```

READ FREQUENCIES, POPULATIONS, WIDTHS

```
9 DO 10 J = 1,NA
READ (5, 15) W(J), P(J), T2W, K
FORMAT (3F10.3, 1I0)
IF (T2W .LE. 0.) T2W = .001
T2(J) = T2W*3.14159
14 WRITE (6,15)W(J),P(J), T2W,K
```

READ RELATIVE TRANSITION PROBABILITIES

```
IF(K) 10,10,16
16 DO 20 L= 1,K
READ (5,25)I,M,PL
25 FORMAT(2I10,F10.7)
WRITE (6,26)I,M,PL
26 FORMAT(35X,2I10,F15.8)
20 B(I,M) = -PL
22 CONTINUE
```

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```

      IF(N(5)) 27,27,25
      DO 24 I = 1, NA
24   Q(I) = P(I)
      CALL MARKOV(A,B,P)
      IF(N(5).GT.2) GO TO 27
      DO 28 I = 1,NA
      IF((ABS(Q(I)-P(I))/Q(I)).GT..005) CALL EXIT
28   CONTINUE

```

READ RATE CONSTANTS ETC.

000004

```

27   IF(N(7).GE.1) GO TO 34
      READ (5,35)R,FI,FF,STEP
35   FORMAT(4F10.4)
      TRACE
34   WRITE (6,36) R,FI,FF,STEP
36   FORMAT(/ 13X,4F15.5 /)
      IF(STEP)380,1,380
80   IF((FF-FI)/STEP)1,1,383
83   IF((FF-FI)/STEP-1800.)38,38,1
38   NO = 0
      DO 82 I = 1,NA
      TR(I) = 0.0
      DO 82 J = 1,NA
82   TR(I) = TR(I) - B(I,J)
      DO 70 I = 1,NA
      DO 70 J = 1,NA
      D(I,J) = 0.
      C(I,J) = B(I,J)*R
      DO 49 I = 1,NA
49   C(I,I) = TR(I)*R + T2(I)
      AMAX = 0.0

```

STEP FREQUENCY

000004

```

43   NO = NO + 1
      ANO = NO-1
      F = FI +ANO*STEP
      IF(F-FF) 41,41,90
41   G = F*6.283185
      DO 40 I = 1,NA
40   D(I,I) = -G + W(I)*6.283185
      CALL INVC(NA,A,C,D)
      S = 0.0
      DO 50 I = 1,NA
      DO 50 J = 1,NA
50   S = S + A(I,J) * P(I)
      SR(NO) = ABS(S) +SR(NO)
      IF(ABS(SR(NO)).GT.AMAX) AMAX = ABS(SR(NO))
55   IF(N(3)) 59,59,43
59   WRITE (6,51) F,SR(NO)
51   FORMAT(20X,F10.3,F25.9)
      FG(NO) = F
      GO TO 43
0    NO = NO - 1
      FS = FS - STEP
89   IF(N(2)) 1,1,57
7    CONTINUE
2    CONTINUE
      GO TO 1
      END

```

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```

SUBROUTINE INVC (N, A, C, D)
INVERSION OF COMPLEX MATRIX WITHOUT COMPLEX ARITHMETIC
USES  A # %C & DC-1D<-1
DIMENSION A(25,25),C(25,25),D(25,25),E(25,25)
DO 10 I = 1,N
DO 10 J = 1,N
10 E(I,J) = C(I,J)
CALL INV(N,E)
DO 20 I = 1,N
DO 20 J = 1,N
A(I,J) = 0.
DO 20 K = 1,N
20 A(I,J) = A(I,J) + D(I,K)*E(K,J)
DO 30 I = 1,N
DO 30 J = 1,N
E(I,J) = 0.
DO 30 K = 1,N
30 E(I,J) = E(I,J) + A(I,K)*D(K,J)
DO 40 I = 1,N
DO 40 J = 1,N
40 A(I,J) = C(I,J) + E(I,J)
CALL INV(N,A)
RETURN
END

```

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```

      SUBROUTINE INV(N,A)
      SINGLE PRECISION REAL MATRIX INVERSION
      DIMENSION A(25,25)
      DIMENSION IPIVOT(25),INDEX(25,2)
      EQUIVALENCE (N1,IROW,JROW),(ICOLUMN,JCOLUMN,JCOLUMN,K1)
15    DO 20 J=1,N
20    IPIVOT(J)=0
30    DO 550 I=1,N
.0    AMAX = 0.
.5    DO 105 J=1,N
.0    IF (IPIVOT(J) - 1) 60, 105, 60
.60   DO 100 K=1,N
.70   IF(IPIVOT(K)-1)80,100,790
.80   IF(ABS(AMAX)-ABS(A(J,K))) 85,100,100
35    IROW=J
.90    ICOLUMN=K
.75    AMAX=A(J,K)
100   CONTINUE
105   CONTINUE
110   IPIVOT(ICOLUMN)=IPIVOT(ICOLUMN)+1
130   IF(IROW-ICOLUMN) 150,260,150
150   DO 200 L=1,N
160   SWAP=A(IROW,L)
170   A(IROW,L)=A(ICOLUMN,L)
200   A(ICOLUMN,L)=SWAP
260   INDEX(I,1)=IROW
270   INDEX(I,2)=ICOLUMN
310   PIVOT=A(ICOLUMN,ICOLUMN)
330   A(ICOLUMN,ICOLUMN)= 1.
340   DO 350 L=1,N
350   A(ICOLUMN,L)=A(ICOLUMN,L)/PIVOT
380   DO550 L1=1,N
390   IF(L1-ICOLUMN) 400,550,400
.00   T=A(L1,ICOLUMN)
.20   A(L1,ICOLUMN)= 0.
.30   DO 450 L=1,N
.50   A(L1,L)=A(L1,L)-A(ICOLUMN,L)*T
.50   CONTINUE
.00   DO710 I=1,N
.10   L=N+1-I
.20   IF(INDEX(L,1)-INDEX(L,2)) 630,710,630
.30   JROW=INDEX(L,1)
.40   JCOLUMN = INDEX(L,2)
.50   DO 705 K=1,N
.60   SWAP=A(K,JROW)
.70   A(K,JROW)=A(K,JCOLUMN)
.00   A(K,JCOLUMN)=SWAP
.05   CONTINUE
.710  CONTINUE
.90   CONTINUE
      RETURN
      END

```

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```

SUBROUTINE MARKOV (A, B, P)
DIMENSION A(25,25), B(25,25), P(25), N(5)
COMMON N
NA = N(1)
DO 50 I = 1, NA
  TR = 0.
  DO 51 J = 1, NA
    1  TR = TR - B(I,J)
  0  B(I, I) = TR
    DO 10 I = 1, NA
      DO 10 J = 1, NA
        0  A(I,J) = -B(I,J)
      DO 20 I = 1, NA
        0  A(I, NA) = 1.
      CALL INV(NA,A)
      DO 30 I = 1, NA
        10  P(I) = A(NA, I)
      WRITE (6, 35) (P(I), I=1,NA)
    35  FORMAT (5X, 10F12.8)
      NM = NA - 1
      DO 40 I = 1, NM
        IP = I + 1
        DO 40 J = IP, NA
          IF (ABS ((B(I,J) * P(I)) - (B(J,I) * P(J)))
            1 + (B(I,J) * P(I)) + (B(J,I) * P(J)) * .02 ) 40, 40, 41
        41  WRITE (6, 42) I, J, P(I), P(J), B(I,J), B(J,I)
        42  FORMAT (12H ERROR IN K , 2I10, 4F12.6)
          IF (N(6) - 2) 43, 40, 43
        43  CALL EXIT
        40  CONTINUE
      DO 60 I = 1, NA
        50  B(I, I) = 0.
      RETURN
    END

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5=LP
NMR